



0040-4039(94)02079-5

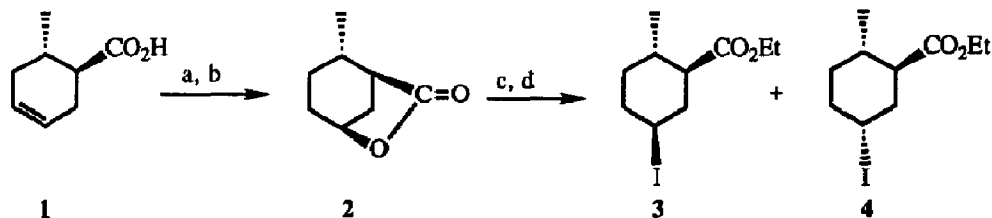
Regioselective Synthesis Of Ceralure B₁ and A, Ethyl *cis*-(and *trans*-) 5-Iodo-*trans*-2-Methylcyclohexane-1-Carboxylate

James W. Avery*, Roy T. Cunningham, and Rolland M. Waters

USDA-ARS-PSI Insect Chemical Ecology Laboratory, Beltsville MD 20705

Abstract: Reaction of *trans*-2-methyl-6-oxabicyclo[3.2.1]octan-7-one with trimethylsilyl iodide, generated in situ, followed by refluxing with ethanol gives regioselectively ethyl *cis*-(and *trans*-) 5-iodo-2-methylcyclohexane-1-carboxylate, a persistent attractant for male Mediterranean fruit flies.

For more than 30 years, trimedlure (TML), a mixture of eight isomers of tert-butyl esters of 4 and 5-chloro-*trans*-2-methylcyclohexane-carboxylic acid (where the *trans* designation refers to the geometric relationship between the vacinal carboxylic ester group and the methyl group), has been widely used as the primary attractant in traps used to monitor and detect male Mediterranean fruit flies.^{1,2,3} Recently, an analog of TML, ceralure (CER), a mixture of eight isomers of ethyl-4(and 5)-iodo-*trans*-2-methylcyclohexane-1-carboxylate, has been found to be a persistent and more active attractant for medflies.⁴ Recent studies^{5,6} have shown that only one of the CER isomers, CER B₁, ethyl-*cis*-5-iodo-*trans*-2-methylcyclohexane-1-carboxylate (3), is the most attractive isomer and it constitutes about 20-25% of commercially synthesized CER. We report here a regioselective synthetic procedure which yields only two CER isomers, CER B₁ and CER A, ethyl-*trans*-5-iodo-*trans*-2-methylcyclohexane-1-carboxylate (4).



Key: a, I₂/KI/NaHCO₃; b, H₂/Pd/EtOH or Bu₃SnH; c, NaI/TMS-Cl/CH₃CN; d, EtOH

The commercial preparation of CER,⁶ adding hydroiodic acid across the olefinic bond of siglure acid, *trans*-2-methylcyclohex-2-ene carboxylic acid (1), followed by subsequent esterification, gives a mixture of eight isomers. Commercial samples of CER consist of a ≥81% mixture of the four *trans* isomers and with not less than 22% CER B₁. To our knowledge, this commercial synthesis of CER is the only preparative method found

in the literature. This research was aimed at development of a synthetic method that would improve the yield of the active CER B₁ isomer. A mixture of epimers of CER B₁ and CER A was prepared as follows: (a) Iodo lactonization⁷ of siglure acid to afford the iodo lactone (*trans*-4-iodo-*trans*-2-methyl-6-oxabicyclo[3.2.1] octan-7-one) (81%), (b) Deiodination of the iodo lactone with tributyltin hydride⁸ or hydrogenolysis using hydrogen over platinum catalyst at 40 psi to give the lactone, *trans*-2-methyl-6-oxabicyclo[3.2.1]octan-7-one (2) (85%); (c) Opening of the lactone^{9,10} by reacting with trimethylsilyl chloride and sodium iodide in acetonitrile, to generate trimethylsilyl iodide in situ, and; (d) Refluxing the intermediate trimethylsilyl ester with ethanol¹¹ to afford the regioselective ethyl *cis*-(and *trans*-) 5-iodo-2-methylcyclohexane-1-carboxylates (3 and 4) (78%, 60:40). The two isomers may be separated by flash chromatography (silica gel, EtOAc/hexane) or semi-preparative HPLC.¹²

A preliminary 2-week field test of the regioselective mixture on cotton wicks indicated competitive fly captures when compared with an equivalent amount of CER B₁ isomer in commercial CER. Because of the 2-fold difference in CER B₁ content, twice as much commercial CER as compared to the new attractant was required to give equal quantities of CER B₁ on the wicks.

In summary, the reaction of 2-methyl-6-oxabicyclo[3.2.1]octan-7-one with trimethylsilyl iodide, generated in situ, followed by refluxing with ethanol gives a 2-2.5 fold increase in the yield of active isomer, CER B₁, than does commercial synthesis. Thus the increased efficacy per dose of material could result in the use of this new attractant as a replacement for TML which is now used at a rate of about 2000 kg per year in the United States.

Acknowledgment: We gratefully acknowledge support in part by a grant from the California Department of Food and Agriculture.

References and Notes

1. McGovern, T. P.; Beroza, M. *J. Org. Chem.* **1988**, *31*, 1472-1477.
2. Sonnet, P. E.; McGovern, T. P.; Cunningham, R. T. *J. Org. Chem.* **1984**, *49*, 4639-4643.
3. Doolittle, R. E.; Cunningham, R. T.; McGovern, T. P.; Sonnet, P. E. **1991**, *J. Chem. Ecol.*, *17*, 475-484.
4. McGovern, T. P.; Cunningham, R. T.; U.S. Patent No. 4 764 366, **1988**.
5. Warthen, J. D. Jr.; Cunningham, R. T.; DeMilo, A. B.; Spencer, S. *J. Chem. Ecol.* **1994**, *20*, 569-578.
6. McGovern, T. P.; Cunningham, R. T.; Foreign Patent No. 7 186 990, **1988**.
7. House, H. O.; Haack, J. L.; McDaniel W. C.; VanDerveer, D. *J. Org. Chem.* **1983**, *48*, 1643-1654.
8. Leibner, J.E.; Jacobus. *J. J. Org. Chem.* **1979**, *44*, 449.
9. Kricheldorf, H. R. *Angew. Chem. Int. Ed. Engl.* **1979**, *18*, 689-690.
10. Olah, G.A.; Narang, S. C.; Balaram Gupta, B.G.; Mehrotra R. *J. Org. Chem* , **1979**, *44*, 1247-1251.
11. Kolb, M.; Barth, J. *Synthetic Comm.* **1981**, *11*, 763-767.
12. Warthen, J. D. Jr.; McGovern, T. P. *Chromatographia* **1990**, *29*, 135-138.

(Received in USA 7 September 1994; revised 12 October 1994; accepted 18 October 1994)